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Synthesis of Homochiral 3-Substituted Glutamic Acids and Prolines from Pyroglutamic Acid

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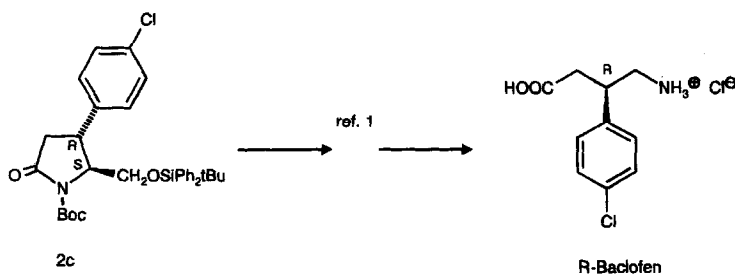
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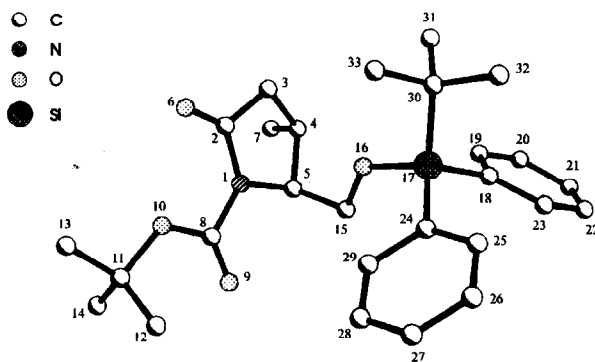
Abstract: Efficient syntheses of (2*S*,3*S*)-methylproline (**5a**) and (2*S*,3*R*)-phenylproline (**5b**) are described, starting from the readily available pyroglutaminol derivatives **2a** and **2b** via conjugate 1,4-addition of organocuprates to **1**. Catalytic hydrogenation of **3** from the least hindered α -side furnishes **6**, which is transformed to (2*S*,3*R*)-methylproline (**7**). 3-Substituted glutamic acids **8c,d** are provided by a four step procedure from **2c,d**.

We recently reported that 1,4-conjugate additions of Grignard-and Gilman-cuprates to pyroglutaminole derivative **1** give **2a-d** in good yields without racemisation of the stereogenic centre in 5-position.¹ Excellent *trans*-selectivity was observed for this reaction. Pyroglutaminole **2c** was transformed to R-Baclofen, a derivative of the inhibitory neurotransmitter GABA (γ -aminobutyric acid)¹ (Scheme 1).



Scheme 1

Herein we report the transformation of pyroglutaminoles **2a,b** to *cis*- and *trans*-3-substituted proline² derivatives **7** and **5a,b**. On the other hand **2c,d** are converted to the 3-substituted glutamic acids³ **8c,d**.



X-ray structure of 2a

For the synthesis of *trans*-3-substituted proline derivatives (Scheme 2), pyroglutamate 2a (the *trans*-configuration was unambiguously assigned by x-ray structure analysis¹⁰) and 2b were reduced with $\text{BH}_3\cdot\text{S}(\text{CH}_3)_2$ to the protected prolinol derivatives 4a,b with 89% and 60% yield respectively. Deprotection of the alcohol function with TBAF provided the N-Boc prolinols in 93% and 86% yield. Sharpless oxidation with $\text{RuCl}_3/\text{NaIO}_4$ ⁴ gave N-Boc prolines 5a,b. After deprotection with TFA, (2S,3S) 3-methylproline (5a)⁹ and (2S,3R)

3-phenylproline (5b)⁹ were obtained in 40% and 31% respectively.

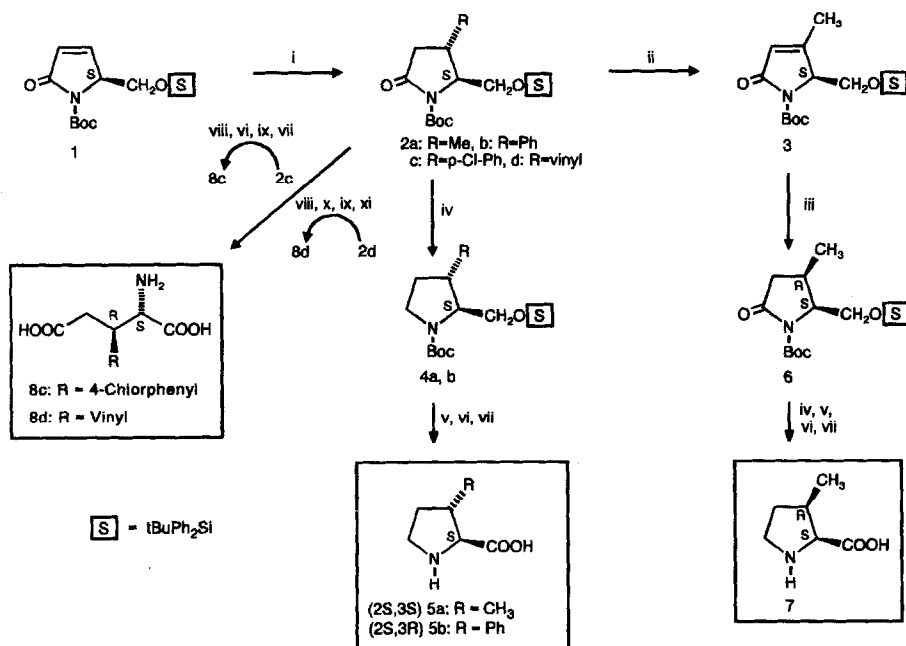
For the synthesis of glutamates 8c,d, 2c,d were deprotected with tetrabutylammonium fluoride in THF/AcOH. The addition of acetic acid was necessary to prevent ring expansion to the δ -lactone derivative.⁵ With HF/ CH_3CN only cleavage of the N-Boc protecting group was observed.

Oxidation of the pyroglutaminol 2c with $\text{RuCl}_3/\text{NaIO}_4$ ⁴ furnished the corresponding pyroglutamic acid derivative, which was ring opened with LiOH/THF⁶ to the N-Boc protected glutamic acid derivative 8c. Deprotection of with TFA provided glutamate 8c⁹ in 42% overall yield from 2c.

As the vinyl group is sensitive to oxidation under the Sharpless conditions the alcohol of 2d was oxidized to the t-butylester with $\text{CrO}_3/\text{Ac}_2\text{O}/t\text{-BuOH}$ ⁷. Extractive isolation of the product was facilitated by both protecting groups. Ring opening reaction according to Grieco's method⁶ and cleavage of the protecting groups with 6M HCl furnished the glutamic acid derivative 8d⁹ in 25% combined yield from 2d.

(2S,3R)-*cis*-3-Methylproline (7) was prepared in a similar reaction sequence as described for 5a and 5b. Starting with 2a, the double bond was introduced via phenylselenylation and oxidative elimination^{21,8,1} to give 3 in 72% yield. Catalytic hydrogenation of 3 in ethyl acetate with Pd/C from the less-hindered α -side of 3 resulted in a single diastereomer 6 (73%). *Trans* isomer 2a could not be detected with ¹H- and ¹³C-NMR spectroscopy in the crude reaction product. After reduction of the amide group and O-deprotection, the alcohol was oxidized⁶ to N-Boc-2S,3R-methylproline. N-Deprotection with TFA provided 7⁹ in 19% overall yield from 3. Pharmacological studies with 8c,d will be published elsewhere in due course.

In summary, the synthesis of enantiopure prolines and 3-substituted glutamic acids (> 98 % ee, in comparison with independently synthesized 5a^{2c}, 5b^{2h}, 7^{2c}) from inexpensive S-pyroglutamic acid has been accomplished. The synthetic strategy outlined in this report would be equally applicable for the synthesis of ent 5a, 5b, 7, 8c, 8d, starting with R-pyroglutamic acid.



Scheme 2

i: $5R_2\text{LiCu}$ or $5R_2\text{MgBrCu}$, $2\text{Me}_3\text{SiCl}$, Et_2O ; ii: 1. HMDS/BuLi , PhSeCl -78°C , 2. H_2O_2 ; iii: Pd/C/H_2 ; iv: $\text{BH}_3\cdot\text{S}(\text{CH}_3)_2$; v: Bu_4NF ; vi: $\text{RuCl}_3/\text{NaIO}_4$, $\text{CH}_3\text{CN}/\text{H}_2\text{O}$; vii: TFA ; viii: $\text{Bu}_4\text{NF}, \text{AcOH}, \text{H}_2\text{O}$; ix: $1\text{M LiOH}/\text{THF}$; x: $\text{CrO}_3/\text{Py}/\text{DMF}$, $\text{Ac}_2\text{O}, t\text{-BuOH}$; xi: 6M HCl .

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References and Notes

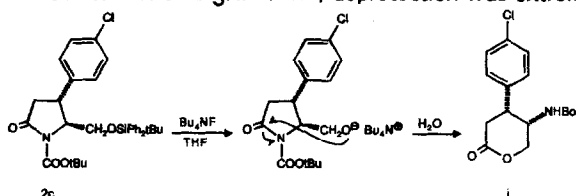
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5. With 2c, the lactonization was accelerated. With 5g, 4H5/THF, the lactonization was extremely slow even if.



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9. 5a: mp.: 245°C (ref. 2c: 240°-249°C).- $[\alpha]_D^{20} = -30$ (c=0.27/H₂O); +5 (c=0.24/0.1 M HCl) (ref. 2c: -30 (c=0.1/H₂O)).- ¹H-NMR (D₂O/d₄-methanol): δ (ppm) = 1.31 (3H, d, J=6.7 Hz, CH₃), 1.70 (1H, dq, J_{3,4a}≈J_{4a,5a}≈J_{4a,5b}=8.5 Hz, J_{gem}=13.0 Hz, 4a-H), 2.24 (1H, m, 4b-H), 2.47 (1H, sept, J=7.2 Hz, 3-H), 3.34 (2H, m, 5-H), 3.65 (1H, d, J_{2,3}=8.1 Hz, 2-H).- 5b: mp.: >250°C (ref. 2b: 242°C.- $[\alpha]_D^{20} = +65$ (c=0.2/1 M HCl) [ref. 2b: +69 (c=0.75/6 M HCl)].- ¹H-NMR (D₂O/D₂SO₄/d₆-acetone): δ (ppm) = 2.16 (1H, m, 4-H), 2.41 (1H, m, 4-H), 3.51 (3H, m, 3-H und 5-H), 4.35 (1H, d, J=9.9 Hz, 2-H), 7.26 (5H, m, Ar-H).- T: mp.: 225°C.- $[\alpha]_D^{20} = -30.5$ (c=0.19/0.1 M HCl).- ¹H-NMR (D₂O/D₂SO₄/d₆-acetone): δ (ppm) = 0.93 (3H, d, J=7.1 Hz, CH₃), 1.70 (1H, sept, J=6.3 Hz, 4-H), 2.16 (1H, m, 4-H), 2.72 (1H, sept, J=7.1 Hz, 3-H), 3.27 (1H, m, 5-H), 3.46 (1H, m, 5-H), 3.65 (1H, d, J_{2,3}=7.7 Hz, 2-H).- 8c: mp.: 182°C.- $[\alpha]_D^{20} = -1$ (c=0.2/1 M HCl).- ¹H-NMR (D₂O/D₂SO₄/d₆-acetone): δ (ppm) = 3.00 (2H, m, 4-H), 3.66 (1H, m, 3-H), 4.34 (1H, d, J=5 Hz, 2-H), 7.17 (2H, d, J_{AB}=8.65 Hz, Ar-H), 7.28 (2H, d, J_{AB}=8.66 Hz, Ar-H).- 8d: mp: 179°C.- $[\alpha]_D^{20} = +13$ (c=0.27/H₂O).- ¹H-NMR (D₂O/d₆-acetone): δ (ppm) = 2.57 (1H, dd, J_{3,4a}=8.7 Hz, J_{gem}=16.4 Hz, 4-H), 2.60 (1H, dd, J_{3,4b}=6.1 Hz, J_{gem}=16.5 Hz, 4-H), 3.03 (1H, m, 3-H), 4.09 (1H, d, J_{2,3}=3.9 Hz, 2-H), 5.23 (2H, m, =CH₂), 5.66 (1H, m, -CH=).

10. X-ray structure determination of 2a: colorless transparent platelets of size 0.5x0.7x0.1 mm, space group P2₁, a=9.831, b=13.088, c=11.213 Å, β=105.2°, z=2.1971 (1863 observed) reflections measured on a Siemens R3m/V diffractometer with CuK_α-radiation. Direct methods (SHELXTL), wR=7.5% (unit weights). Atomic distances and angles as well as coordinates are deposited at the Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, UK.